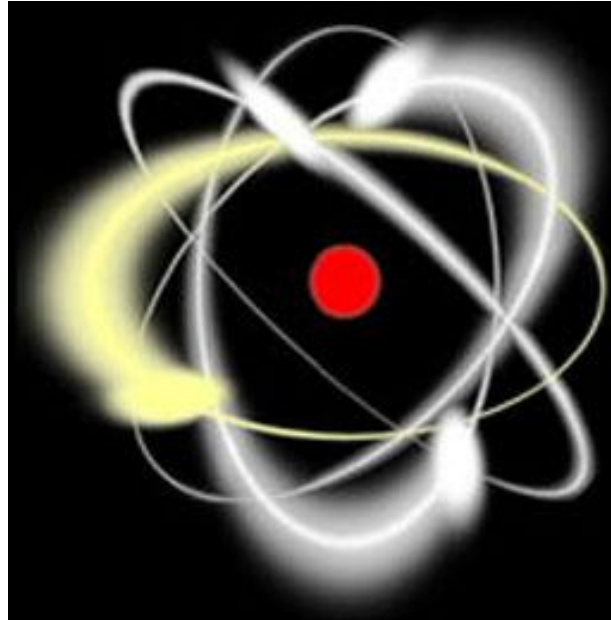


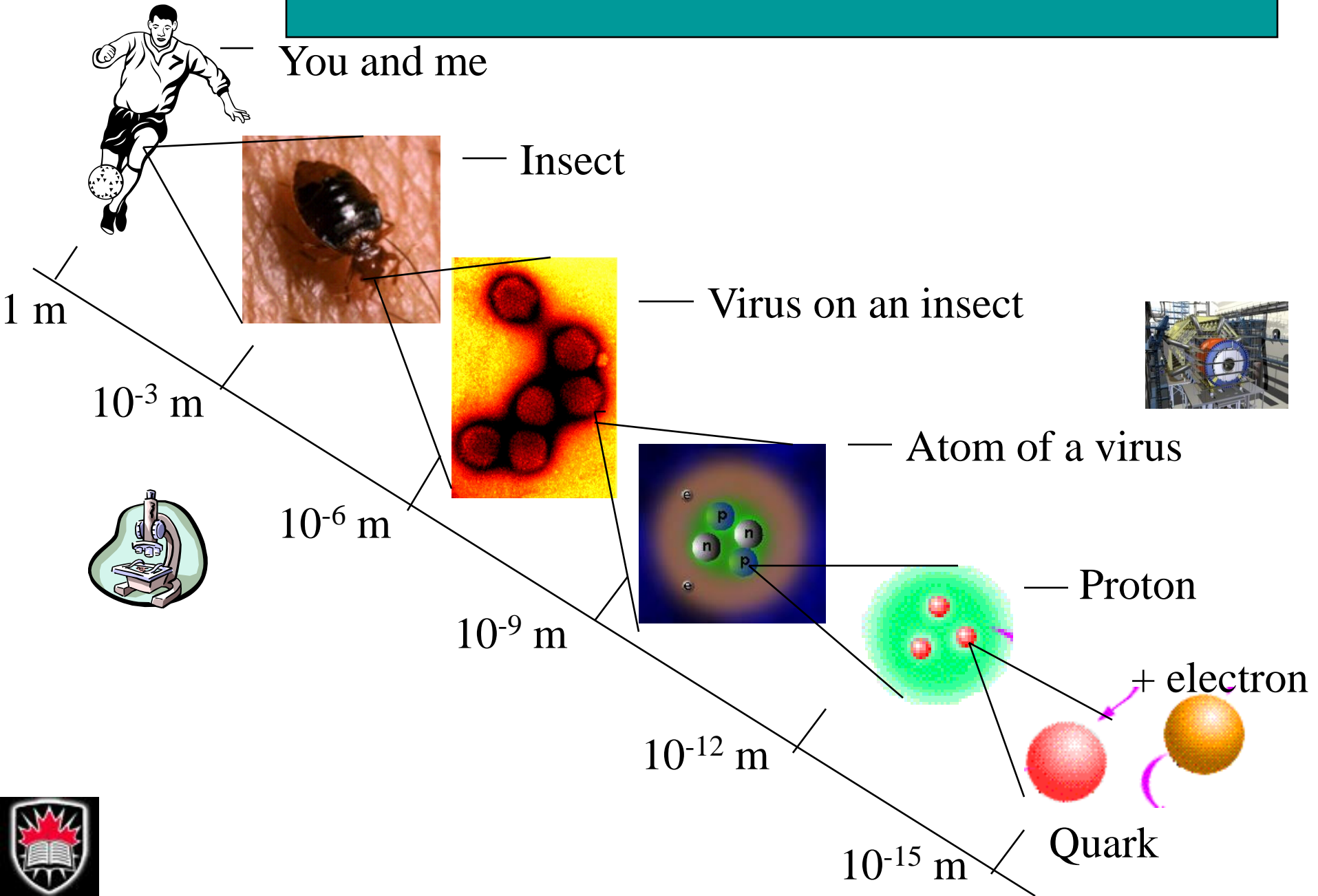
Measurement of Charge to Mass Ratio For an Electron



(Thomson's Experiment)



The scale of the subatomic world



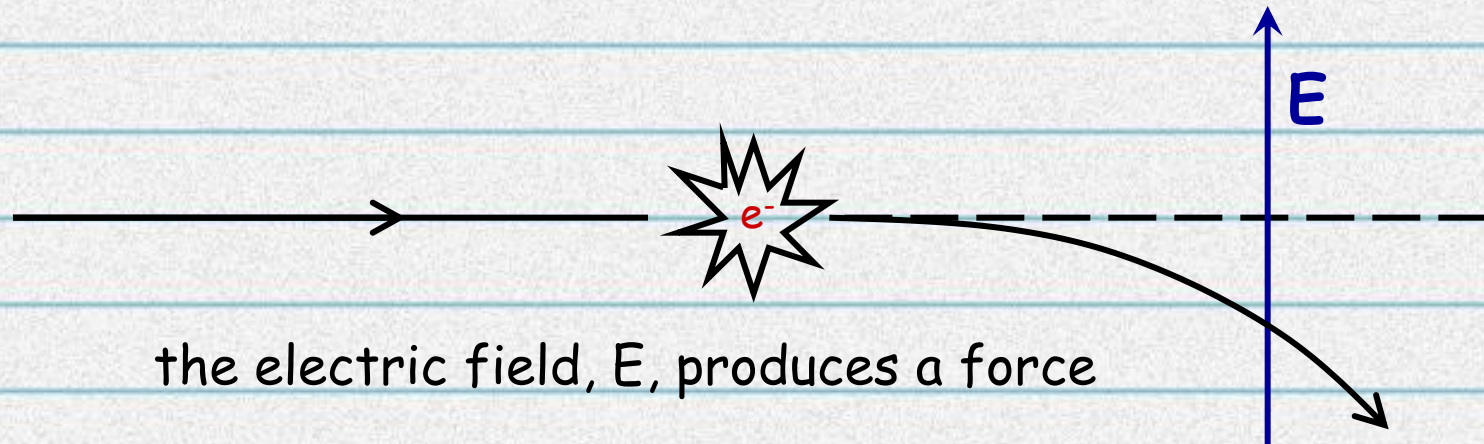
The Electron:

- is an elementary particle: smallest speck of matter
- is normally found in the immediate vicinity of a nucleus, forming an atom
 - Mass (m_e): $9.11 \times 10^{-31} \text{ kg}$
 - Charge (e): $1.6 \times 10^{-19} \text{ C}$ ($\text{C} = \text{Coulombs}$)
- Charge is found by Millikan's Oil Drop experiment
- So, if we can find e/m_e , we can determine m_e
- In 1897, J.J. Thomson found this value
 - Ratio (e/m_e): $-1.76 \times 10^{11} \text{ C/kg}$
- Your Job: try to repeat that measurement today



Forces affecting the electron:

1. The electric field:



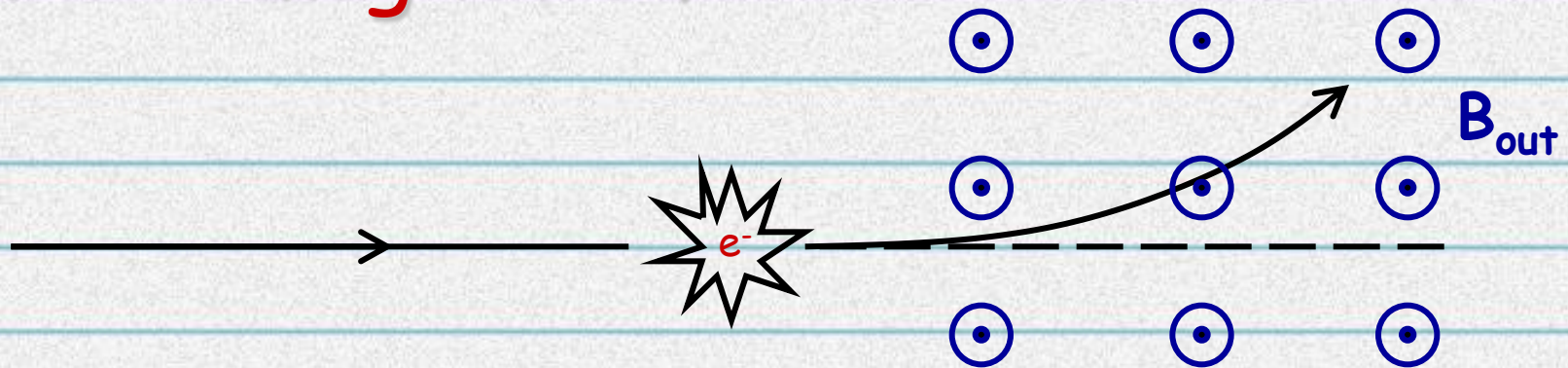
the electric field, E , produces a force

$$F_E = q \cdot E \quad (q = e, \text{ the charge of the electron})$$

The electric field, E , always points in the direction that a +ve charge would move if it were within the field.

Forces affecting the electron:


2. The magnetic field:



The magnetic field (B_{out}) produces a force:

$$F_B = B_{out} \cdot e \cdot v \quad (v \text{ is the velocity of the electron})$$

This force is perpendicular to both B_{out} and v .

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Finding e/m :

- Electrons move in circles in magnetic fields
- This motion produces a **centripetal force**

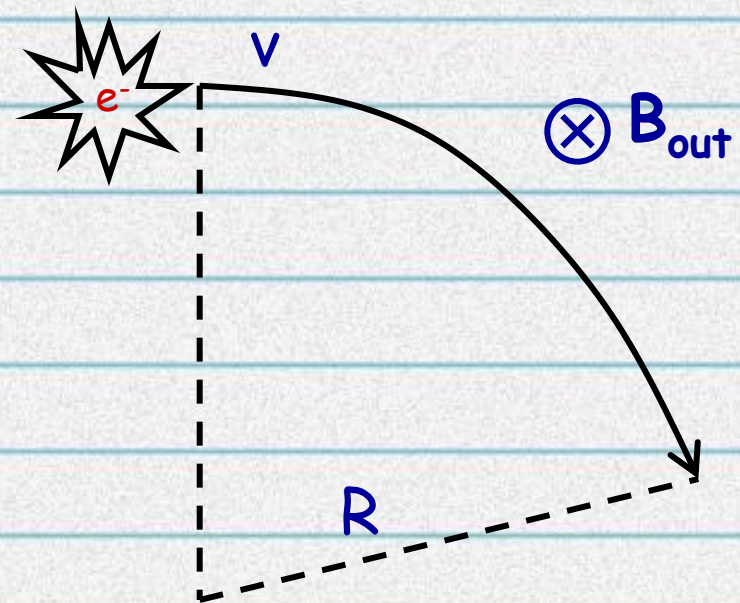
$$F = \frac{mv^2}{R}$$

- We can equate this to F_B :

$$Bev = \frac{mv^2}{R}$$

- Re-arranging:

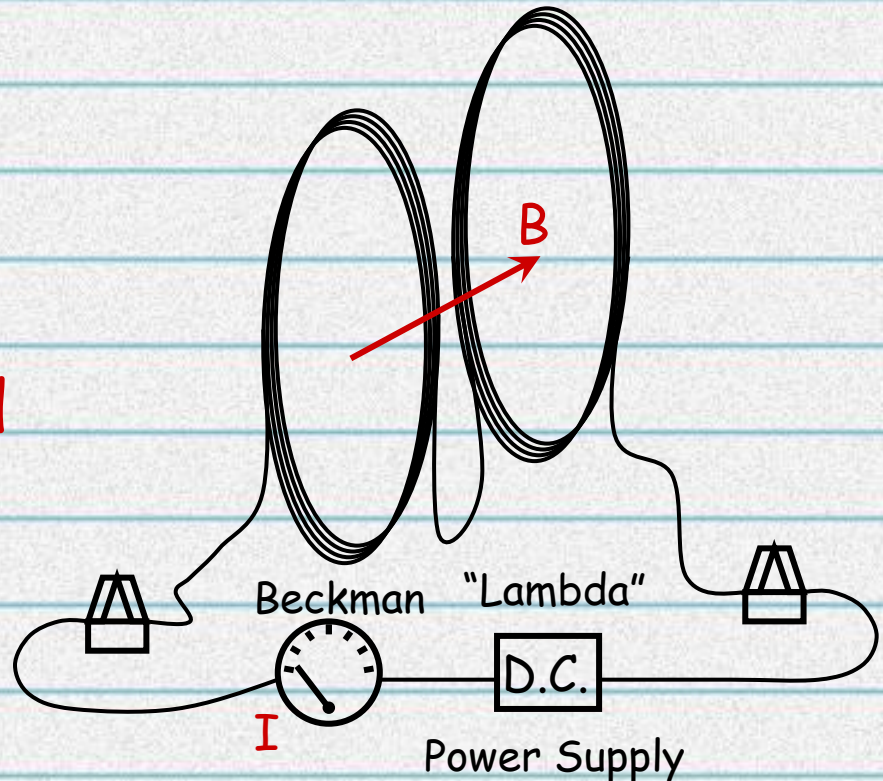
$$\frac{e}{m} = \frac{v}{BR}$$



- Thus, to find e/m we need to know 3 things:
the magnetic field, B , the radius of curvature, R , and
the velocity of the electrons, v .

Finding the magnetic field, B:

- "Helmholtz" coil arrangement delivers uniform magnetic field

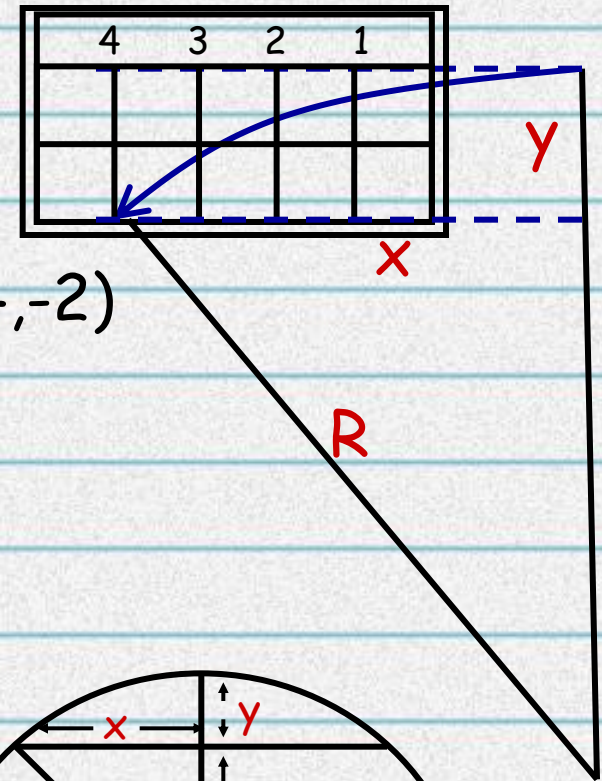


- B depends on current, and is calibrated to be:
$$B = I \times 4.23 \times 10^{-3} \text{ Wb m}^{-2}$$

(I measured in Amps)

Determine the radius, R:

- Measure y - deflection at a distance, x , from the exit of the electron



$$(x, y) = (4, -2)$$

- Given (x, y) - find R

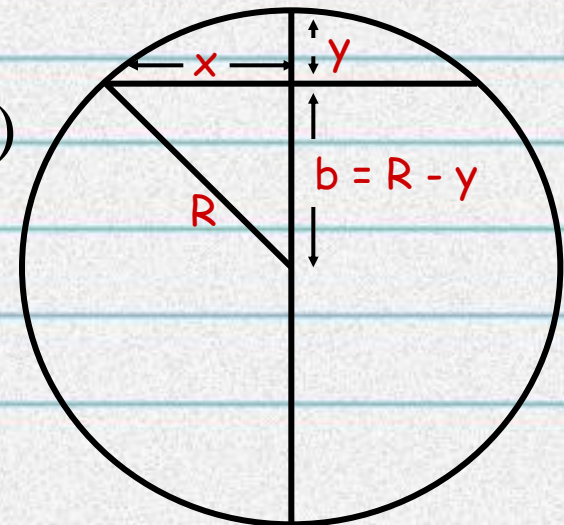
$$R = \frac{x^2 + y^2}{2y}$$

- To prove this, try Pythagoras:

$$(R - y)^2 = b^2 = (R^2 - x^2)$$

$$R^2 - 2Ry + y^2 = R^2 - x^2$$

$$R = \frac{x^2 + y^2}{2y}$$



Determine the velocity, v :

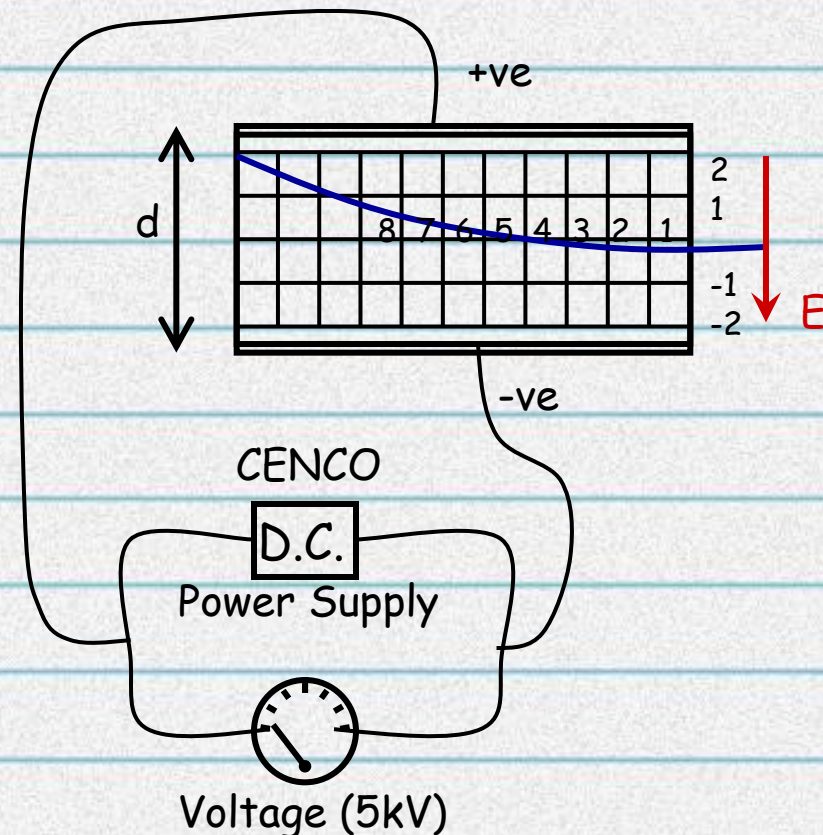
- Now switch on the **Electric Field**
 - Use it to **cancel** the effects of the **B field**

$$E = \frac{V}{d} = \frac{V}{0.052}$$

- Use the Electric field to cancel the deflection at the **x value** where the magnetic deflection was measured

$$Ee = Bev$$

$$v = \frac{E}{B} \quad (\text{m/s})$$



Procedure:

- Plug in transformer for cathode heater supply
- Switch on CENCO D.C. supply to accelerate electrons down the tube and onto the screen
set to $\sim 2400\text{ V}$
- There should now be a blue trace on the screen
- Switch on Beckman meter and Lambda power supply unit connected to coils
set to $\sim 0.2\text{ A}$
- Determine B



Procedure:

- Measure (x,y) :
determine R
- Switch on second CENCO DC supply connected to electrodes on top of mica screen
set to ~ 1.5 kV
- Adjust to cancel deflection:
determine v
- Fill in worksheet and calculate e/m
- Switch off all power supplies



	<i>Thomson's e/m experiment</i>								
		Your Measurements		Units	Calculated quantities			Units	
Determining the		χ		metres	$R = (\chi^2 + y^2) / 2y$			metres	
Radius(R) of the track		y		metres					
Determining the		I		amps	$B = 4.23 \times 10^{-3} I$			Tesla	
Magnetic Field (B)									
Determining the		V		volts	$E = V / 0.052$			volts/metre	
Electric Field (E)									
Determining the					$v = E / B$			metres/sec	
electron's velocity(v)									

